A MONTHLY, OPEN ACCESS, PEER REVIEWED (REFEREED) INTERNATIONAL JOURNAL Vol. 02, Issue 03, March 2023

# The role of Nanoscience in pollution detection and control

## Shalini Singh<sup>1</sup>

<sup>1</sup>Department of Zoology, Govt. Degree College, Kuchlai, Sitapur, Uttar Pradesh

Received: 15 March 2023 Accepted & Reviewed: 25 March 2023, Published: 31 March 2023

## **Abstract**

Nanoscience plays a pivotal role in addressing the growing global challenge of pollution by enabling highly sensitive detection and efficient control of environmental contaminants. With the development of nanomaterials such as carbon nanotubes, metal oxide nanoparticles, quantum dots, and nanosensors, it has become possible to detect pollutants like heavy metals, pesticides, and toxic gases at trace levels. These nanostructures exhibit unique physicochemical properties—including high surface area, reactivity, and selectivity—that enhance the sensitivity and accuracy of pollutant monitoring systems. Moreover, nanotechnology contributes to pollution control through advanced filtration systems, photocatalytic degradation of pollutants, and green remediation technologies. The integration of nanoscience into environmental monitoring and waste management represents a transformative step toward sustainable and cleaner ecosystems. This paper explores the mechanisms, applications, and future prospects of nanoscience in pollution detection and control.

**Keywords:-** Nanoscience, Environmental Pollution, Nanosensors, Pollution Detection, Nanomaterials, Photocatalysis, Environmental Remediation, Green Technology

# Introduction

Human activity and industrialization leads to the environment filled with different kinds of pollutants. Air is filled with cabonmonooxide (CO), Cholofluorocarbons (CFC), heavy metals (lead, arsenic, chromium, cadmium, mercury, zinc), hydrocarbons, nitrogen oxide, sulphur oxide and particulates. Water pollution is caused by numerous factors including sewage, oils spills, leaking of fertilizers, herbicides and pesticides from land etc. Contaminants are most often measured in parts per million (ppm) or parts per billion (ppb) and their toxicity defined by a 'toxic level'. The toxic level for arsenic, for instance is 10 ppm in soil whereas for mercury is 0.002 ppm in water. Therefore, very low concentrations of a specific contaminant can be toxic. There is need for technologies are capable of monitoring recognizing and ideally treating such small amount of contaminants in air, water and soil. Environmental nanotechnology is considered to play a key role in the shaping of current environmental engineering and science. The nanotechnological applications and products can lead to a cleaner and healthier environment [1]. Maintaining and re-improving the quality of water, air and soil, so that the Earth will be able to support human and other life sustainably, are one of the great challenges of our time. Nanotechnology can play a vital role in providing clean air, water and soil in an efficient and cheap way [2].

Nanoscience allows designing and manipulating materials at the atomic and molecular level. Nanomaterials can be fabricated with specific properties that can recognize a particular pollutant within a mixture. The small size of nonmaterials together with their high surface to volume ratio can lead to very sensitive detection. These properties will allow developing highly miniaturize, accurate and sensitive pollution-monitoring devices ('nano-sensors'). Nanomaterials can also be engineered to actively interact with a pollutant and decompose it in less toxic species. Thus, in the future nanotechnology could be used not only for detecting contaminated sites but also treating them.

A MONTHLY, OPEN ACCESS, PEER REVIEWED (REFEREED) INTERNATIONAL JOURNAL Vol. 02, Issue 03, March 2023

### **Pollution Detection And Sensing**

Fortification of the human health and protection of the environment requires the rapid, sensitive detection of pollutants and pathogens with molecular precision. Sensors are needed for in situ detection, as miniaturized portable devices, and as remote sensors, for the real-time monitoring of large areas in the field. A sensor is a device built to detect a specific biological or chemical compound, usually producing a digital electronic signal upon detection. Sensors are now used for the identification of toxic chemical compounds at ultra low levels (ppm and ppb) in industrial products, chemical substances, water, air and soil samples, or in biological systems. Nanotechnology can improve current sensing technology in various ways. First, by using nanomaterials with specific chemical and biological properties, the sensor selectivity can be improved, thus allowing isolating a specific chemical or biological compound with little interference. Hence, the accuracy of the sensors is improved. As with other nano-engineered products discussed in this document, the high surfaceto-volume ratio of nanomaterials increases the surface area available for detection, which in turn has a positive effect on the limit of detection of the sensor, therefore improving the sensitivity of the device. Scaling down using nanomaterials allows packing more detection sites in the same device, thus allowing the detection of multiple analytes. This scaling-down capability, together with the high specificity of the detection sites obtainable using nanotechnology, will allow the fabrication of super-small 'multiplex' sensors, this way lowering the cost of the analysis and reduce the number of devices needed to perform the analysis with an economic benefit. Advancements in the field of nanoelectronics will also allow the fabrication of nanosensors capable of continuous, real time monitoring [3].

Various nanostructured materials have been explored for their use in sensors for the detection of different compounds [4]. An example is silver nanoparticle array membranes that can be used as flow-through Raman scattering sensors for water quality monitoring [5]. The particular properties of carbon nanotubes (CNTs) make them very attractive for the fabrication of nanoscale chemical sensors and especially for electrochemical sensors [6-9]. A majority of sensors described so far use CNTs as a building block. Upon exposure to gases such as NO2, NH3 or O3, the electrical resistance of CNTs changes dramatically, induced by charge transfer with the gas molecules or due to physical adsorption [10, 11]. The possibility of a bottomup approach makes the fabrication compatible with silicon microfabrication processes. The sensor is made of an array of electrode pairs fabricated on a silicon chip and separated by few nanometres. When the electrodes are exposed to a solution of water containing metal ions, these deposit inside the nano-gap in between the electrodes. Once the deposited metal bridges the gap a 'jump' in conductance between the electrodes is registered. The size of the gap, being only few nanometres, allows the detection of a very low concentration of metal ions. This type of sensor is called 'nanocontact sensor'. [12]. The connection of CNTs with enzymes establishes a fast electron transfer from the active site of the enzyme through the CNT to an electrode, in many cases enhancing the electrochemical activity of the biomolecules [8]. In order to take advantage of the properties of CNTs, they need to be properly functionalized and immobilized. CNT sensors have been developed for glucose, ethanol, sulfide and sequence-specific DNA analysis [8]. Trace analysis of organic compounds, e.g. for the drug fluphenazine, has also been reported [13]. Nanoimmunomagnetic labeling using magnetic nanoparticles coated with antibodies specific to a target bacterium have been shown to be useful for the rapid detection of bacteria in complex matrices [14]. Materials that are more environment-friendly fabricated using nanotechnology include biodegradable elf-cleaning glasses, such as Activ<sup>TM</sup> Glass [15], the glass is composed of a special coating made of nanocrystals of TiO<sub>2</sub> which, once exposed to daylight, reacts in two ways. First, it breaks down any organic dirt deposits on the glass and second, when exposed to water, it allows rain to 'sheet' down the glass easily and washes the loosened dirt away.

A MONTHLY, OPEN ACCESS, PEER REVIEWED (REFEREED) INTERNATIONAL JOURNAL Vol. 02, Issue 03, March 2023

#### **Nanocatalsis**

A catalyst is a substance that increases a chemical reaction rate without being consumed or chemically altered. One of the most important properties of a catalyst is its 'active surface' where the reaction takes place. The 'active surface' increases when the size of the catalysts is decreased. The higher is the catalysts active surface, the greater is the reaction efficiency. Also, research has shown that the spatial organization of the active sites in a catalyst is important as well [16]. Both properties (nanoparticle size and molecular structure/distribution) can be controlled using nanotechnology. In the environmental field, nanocatalysis is being investigated for **desulphurizing fuels**, with the aim of developing 'clean' fuels containing very low sulphur products (produced in the fuel during its refining process and responsible for generating sulphuric acid upon fuel combustion).

Another area where nanotechnology is making a contribution is the development of **fertilizers** and **wood treatment** products that are more stable and leach less into the environment. For instance, researchers at the Michigan State University have incorporated biocides for wood treatment inside polymeric nanoparticles. Their small size allows them to efficiently travel inside the very fine, sieve-like structure of wood. At the same time, the biocide, being safely trapped inside a 'nanoshell', is protected from leach and random degradative processes [17].

### **Green Manufacturing**

Manufacturing processes are always accompanied by the production of diverse waste products, many of which pose a threat to the environment and thus need to be removed and treated. Green manufacturing includes the development of new chemical and industrial procedures (for instance water-based rather the solvent-based processes); reduction in the use of unsafe compounds (such as metals); development of 'green' chemicals that are more environment-compatible; and efficient use of energy. In terms of its application to the reduction of manufacturing waste, nanotechnology can contribute in two ways: by directing the manufacturing to be more controlled and efficient, and by using nanomaterials (such as catalysts) that can raise the manufacturing efficiency while reducing or eliminating the use of toxic materials. Overall, nanotechnology has the potential of making industrial processes more efficient in terms of energy usage and material usage, while minimizing the production of toxic wastes. The application of 'green nanotechnology [18] to manufacturing includes bottom-up, atomic-level synthesis for developing improved catalysts; inserting information into molecules to build new materials (such as DNA) through highly specific synthetic routes; scaling down material usage during chemical reaction by using nanoscale reactors; and improving manufacturing to require less energy and less toxic materials.

An example of 'green nanotechnology' is the development of aqueous-based microemulsions to be used in alternative to volatile organic compounds (VOCs) in the cleaning industry. These toxic and potentially carcinogenic compounds, such as chloroform, hexane, percholoroethylene, are conventionally used in the cleaning industry (like the textile industry) as well as in the oil extraction industry. Microemulsions contain nano-sized aggregates that can be used as 'receptors' for extracting specific molecules at a nanoscale level. Other example is microemulsions having water-attractive and water-repellent "linkers" inserted between the head and tail parts of a surfactant molecule [19]. The result is a surfactant that has a very low interfacial tension with a wide range of oils. When tested for cleaning textiles from motor oil residues, as well as for extracting edible oil from oilseeds, the microemulsions were found to be very competitive with conventionally used VOCs, both in terms of extraction yield and simplicity of the process.

Nanowires of semiconductors such as silicon has established knowledge for the chemical modification

A MONTHLY, OPEN ACCESS, PEER REVIEWED (REFEREED) INTERNATIONAL JOURNAL Vol. 02, Issue 03, March 2023

of their surface. Boron-doped **silicon nanowires** (SiNWs) have been used for the sensitive real-time electrical detection of proteins, antibodies the metabolic indicator calcium[20, 21] and glucose in water [22]. The small size and the capability of these semiconductor nanowires to detect in real-time a wide range of analytes could be used for developing sensors for detecting pathogens, chemical and biological agents in water, air and food.

Nanotechnology's potential and promise have steadily been growing throughout the years. The world is quickly accepting and adapting to this new addition to the scientific toolbox. Although there are many obstacles to overcome in implementing this technology for pollution control, science is constantly refining, developing, and making breakthroughs.

#### References-

- 1. Masciangioli T. and Zhang W. X. Environmental technologies at the nanoscale, *Environ. Sci. Technol.* **37(5)**, 102A-108A **(2003)**
- 2. Hillie T., Munasinghe M., Hlope M. and Deraniyagala Y. Nanotechnology, Water and Development, Meridian Institute (2006)
- 3. Report from Applications of Nanotechnology: Environment-Luisa Filipponi & Duncan Sutherland, Nanocap (2007)
- 4. Vaseashta A., Vaclavikova M., Vaseashta S., Gallios G., Roy P. and Pummakarnchana O. Nanostructures in environmental pollution detection, monitoring, and remediation, *Sci. Technol. Adv. Mater.* **8** (1), 47-59 (2007)
- 5. Taurozzi, J. S. and Tarabara, V. V. Silver nanoparticle arrays on track etch membrane support as flow-through optical sensors for water quality control, *Environ. Eng. Sci.* **24 (1)**, 122-137 **(2007)**
- 6. Wang J. Carbon-Nanotube Based Electrochemical Biosensors: A Review, *Electroanalysis* 17 (1), 7-14 (2005)
- 7. Trojanowicz M. Analytical applications of carbon nanotubes: A review, *Trends Anal. Chem.* 25 (5), 480-489 (2006)
- 8. Valcarcel M. Simonet B.M., Cardenas S. and Suarez B. Present and future applications of carbon nanotubes to analytical science, *Anal. Bioanal. Chem.* **382 (8)**, 1783-1790 **(2005)**
- 9. Merkoci A. Carbon Nanotubes in Analytical Sciences, Microchim. Acta 152 (3), 157-174(2006)
- 10. Dai L., Soundarrajan P. and Kim T. Sensors and sensor arrays based on conjugated polymers and carbon nanotubes, *Pure Appl. Chem.* **74 (9)**, 1753-1772 **(2002)**
- 11. Sano N. and Ohtsuki F. Carbon nanohorn sensor to detect ozone in water, *J. Electrostat.* **65 (4)**, 263-268 **(2007)**
- 12. Li J., Koehne J.E., Cassell A.M., Chen H., Ng H.T., Ye Q., Fan W, Han J. and Meyyappan M. Inlaid Multi-Walled Carbon Nanotube Nanoelectrode Arrays for Electroanalysis, *Electroanalysis*, **17(1)**, 15-27 **(2005)**
- 13. Zeng B.Z. and Huang F. Electrochemical behavior and determination of fluphenazine at multi-walled carbon nanotubes/(3-mercaptopropyl) trimethoxysilane bilayer modified gold electrodes, *Talanta* **64 (2)**, 380-386 **(2004)**

A MONTHLY, OPEN ACCESS, PEER REVIEWED (REFEREED) INTERNATIONAL JOURNAL Vol. 02, Issue 03, March 2023

- 14. Chang S. C. and Adriaens P. Nano-immunodetection and quantification of mycobacteria in metalworking fluids, *Environ. Eng. Sci.* 24 (1), 58-72 (2007)
- 15. Activ<sup>™</sup> Glass, Pilkington, www.pilkington.com
- 16. Gemming S. and Seifert G. Catalysts on the edge, *Nature* 2, 21-22 (2007)
- 17. Liu Y., Yan L., Heiden P. and Laks P. Use of nanoparticles for controlled release of biocides in solid wood', J. Appl. Poly. Sci. 79 (3), 458-465 (2001)
- 18.For a review on 'Green nanotechnology' covering definition, concepts and applications see K. Schmidt, Green Nanotechnology (PEN8), Woodrow Wilson Center International Center for Scholars, free to download from www.nanotechproject.org/reports.
- 19. Acosta E.J., Nguyen T., Witthayapanyanon A., Harwell J. H., and Sabatini D.A. Linker-based biocompatible microemulsions, *Environ. Sci. Tecnol.* **39** (5), 1275-1282 (2005)
- 20. Cui Y., Park H. and Lieber C.M. Nanowire nanosensors for highly sensitive and selective detection of biological and chemical species, *Science* **293** (5533), 1289-1292 (2001)
- 21. Patolsky F. and LieberC.M. Nanowire nanosensors, *Materials Today* 8 (5) 20-28(2005)
- 22 Shao M., Shan Y, Wong N. and Lee S. Silicon nanowire sensors for bioanalytical applications: glucose and hydrogen peroxide detection, *Adv. Func. Mater.* **15 (9)**, 1478-1482 **(2005)**